CIA-RDP86-00513R001962210016-3 "APPROVED FOR RELEASE: 09/01/2001

USSR/Cultivated Plants - Commercial. Oil-Bearing. Sugar-Bearing. M

: Rof Zhur Biol., No 13, 1958, 82433 Abs Jour

Protosov, P.V., Yarovenko, G.I.

Some Data on the Influence of Antiseptics on Cotton Author Inst

Title Yield.

Sots. s. kh. Uzbekistana, 1956, No 6, 71-73 Oric Pub

In 1955, laboratory experiments were carried out at the Abstract

Central Station of Fertilizers and Agricultural Soil Science of the All-Union Cotton Scientific Research Instit te for the purpose of a comparative study of the influence of Nts (as all antiseptic) on the dynamics of the formation of nitrate and ammoni in N in sierozens. Expe. riments were conducted in Petri dishes. 100 grams of the soil and 20 milligrams of N in the form of Nag, Na and

Nts were placed in each dish, Assantiseptics, 12% hexachlorane dist (20 milligrams to a dish) and Nts

Card 1/2

- 87 -

USSR/Cultivated Plants - Commercial. Oil-Ecaring. Sugar-Bearing. M
Abs Jour : Ref Zhur Biol., No 18, 1958, 82433

(5 milligrams to a dish) were applied. The resulting data show that an addition to the fertilizers of a small quantity of Nts or hexachlorane is accompanied with an accumulation of armonium N and a slower acidification of it to nitrates. The field test conducted in the same year at Sverdlov Kolkhoz showed that an addition to Naa of antiseptics (hexachlorane, granosan / ethylmer-curochloride /, paraform) increased the cotton wool yield on an average by 3 centuers/ha. The most positive effect on the cotton wool yield was produced by the 12% hexachlorane dust. -- V.F. Nepomiluyev

Card 2/2

USSR / Cultivated Plants. Commercial. Oil-Bearing. Sugar-Bearing.

Abs Jour: Ref Zhur-Biol., No 6, 1958, 25124

: Yarovenko, G.I. Author

: The All-Union Cotton S.R.I. Inst

: A Contribution to the Problem of the Significance Title of Nitrate and Ammonia Nitrogen in Cotton Nourish-

ment

Card 1/2

Orig Pub: Dokl. AN UzbSSR, 1956, No 9, 43-46 (Res. Uzb.)

Abstract: By the isolated feeding method in water and soils cultures the periodic feeding of cotton with nitrate and ammonia nitrogen was studied at the Central Station of Fertilizers and Agricultural Soil Science of the All-Union Cotton Scientific Research Institute. The variant appeared best in the number of bolls having formed and the raw cotton yield, where

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CIA-RDP86-00513R001962210016-3" **APPROVED FOR RELEASE: 09/01/2001**

USSR / Cultivated Plants. Commercial. Oil-Bearing. M-5 Sugar-Bearing.

Abs Jour: Ref Zhur-Biol., No 6, 1958, 25124

Abstract: nitrogen was applied in ammonia form up to the beginning of flowering, and afterwards in nitrate form. Cotton had a higher percentage of pest damage throughout the entire period of vegetation in the presence of nitrate feeding. With ammonia feeding the cotton was somewhat smaller in growth, had dark green leaves and a very low percentage of damage. Cotton's absorption of nitrates and ammonia from nutrient mixtures of Ca (NO₃)₂ and (NH₄)₂SO₄ took place approximately in equal quantities. -- A.M. Shchepetil'nikova

Card 2/2

MAUNEMAN Culvivated Plants. Cormorcial. Oleiferous. Consideration of the Sugar-Rearing Ther That Michael ya No. 5, 1959, 20, 20393 AMS. JOUR : Madraimov, I.M.: Popova, I.M.: Popov, G.P. * AUTHOR INST. A3 Uzbek SSR Production Experiments in Applying Liquid TITLE Nitrogen Fertilizers under Cotton in 1956. V sb.: Ref. naushno-issled. rabot po khlop-kovodstvu. Tashkent, AN UZSSR, 1957, 156-179 ORIG. PUB.: ABSTRACT : Comparative study of liquid ammonia and ammoniate (A) in the kolkhozes of Uzbek SSR in 1956 on different soils showed them to be equally effective. In a number of laboratory tests the volatility of A under varying soil moisture and planting depths, its percolotion with the water flow and the rate of nitrification in the soil. -. D.B. Vakhmistrov * Yarovenko, G.I CARD: 1/1

Country : USSR

.T

Category: Soil Science. Physical and Chemical Properties of Soil.

Abs Jour: RZhBiol., No 18, 1958, No 82095

Author : Yarovenko G. I.

Inst , AS Uzbek SSR

Title : Influence of Irrigation Water on the Movement of Ammonium

Nitrogen in Sicrozen Soil.

Orig Pub: Dok. AN UZSSR, 1957, No 5, 45-48

Abstract: In experiments of the Central Station of Fertilization and Agricultural-Soil Science of the All-Union Cotton Scientific Research Institute in an irrigation area of the republics of Central Asia nitrates were washed to depths of 50 cm in medium loamy sierozem. The distribution of the liquid ammonia depended on

Card : 1/2

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001962210016-3"

Country : USSR

Category: Soil Science. Physical and Chemical Properties of Soil.

Abs Jour: RZhBiol., No 18, 1958, No 82095

pre-existing moisture of the soil. When the moisture of the soil was 5-18% of the weight of the air-dried soil, ammonia diffused in a radius of 5 - 9 cm, with the moisture 22% - 3 cm. Without irrigation ammonia was concentrated at the site of the introduced fertilizer (horizon 0 - 3 cm). The fortilizers applied were: ammonium nitrate, ammonium sulfate, ammonium bicarbonate, ammonia brand A, and liquid ammonia. -- S. A. Nikitin

USSR / Cultivated Plants. Plants for Technical Use. M 011 Plants. Sugar Plants.

Abs Jour : Ref Zhur - Biologiya, No 6, 1959, No. 24965

Author

Inst

: Yarovenko, G. I. : Acadery of Sciences UzSSR

Title

: Some Data on the Effectiveness of Urea on

Cotton-Plant Sowings

Orig Pub

: Dokl. AN UZSSR, 1957, No 8, 47-50

Abstract

: In a field experiment by the Central Station of Fertilizers and Agricultural Soil Science, SoyuzNIKhI [All-Union Scientific Research Institute of Agriculture] (1956), the effectiveness of Ny in the capacity of a nitrogen fertilizer at the pre-sowing application exceeded, and with additional application under the plant equalled the effectiveness of

Card 1/2

CIA-RDP86-00513R001962210016-3 APPROVED FOR RELEASE: 09/01/2001

USSR / Cultivated Plants. Plants for Technical Use. Oil Plants. Sugar Plants.

 \mathbf{M}

: Ref Zhur - Biologiya, No 6, 1959, No. 24965 Abs Jour

> $N_{\text{aa}}\, .$ The speed of nitrification of N_{M} in the sierozem soil equalled the nitrification speed of Naa.

Card 2/2

119

COUNTRY : USSR M-7
GAT #3GRY :

ABS. JOUR. : RZB101., No. /% 1959, No. 87141

AUTHOR : Protasov, P.; Yarovenko, G.

INST. : Use of Calcium Cyanamide as a Nitrogen

Fertilizer for Cotton.

ORIG. PUB.: Khlopkovodstvo, 1957, No 10, 23-26

ABSTRACT: On prolonged storage N_c loses a part of N and becomes unsuited for defcliation of cotton (when the content of N is less than 16%). In this connection the Central Station of Fertilizers and Agricultural Soil Science of Union-NIKhI has conducted in 1946-1957 expendents on utilization of low-N content N_c as fertilizer for cotton. The experiments showed that effectiveness of N_c as a fertilizer depends on the time of its application. Early preplanting application of N_c at the time of autumn- or preplanting plowing eliminates its toxicity to plants and promotes conversion of N to readily assimilable form. Rate of application is 300-400 kg/hectare. This amount of CARD: 1/2

Country CATEGORY USSR M = 7A33. JOUR. : RZBiol., No. /9, 1959, No. 87141 AUTHOR INST. TITLE

ORIG. PUB. :

ABSTRACT: N_c makes it possible to increase the yield by at least 2-3 centners/hectare. In the Uzbek SSR alone, 30-50 thousand hectares can be fertilized by making use of N_c that is not suitable for defoliation. A. M. Smirnov.

CARD: 2/2

CIA-RDP86-00513R001962210016-3" **APPROVED FOR RELEASE: 09/01/2001**

USSR/Cultivated Plants - Commercial. Oil-Bearing. Sugar-Bearing. M

Abs Jour : Ref Zhur Biol., No 18, 1958, 82429

Author : Yaroverko, G.I.

Inst : Academy of Sciences Uzbek SSR

Title : On Testing Bicarbonate of Ammonium on Cotton Plantings.

Orig Pub : Uzssk Fanlar Akad. dokladlari, Dokl. AN Uzssk, 1957,

No 11, 45-47

Abstract : Results of field and laboratory investigations carried

out in 1956 at the Central Station of Fertilizers and Agricultural Science of the All-Union Cotton Scientific Research Institute on the study of amonium bicarbonate prepared by Moscow Institute of Chemical Mechanical Engineering. The field tests were conducted in Tashkentskaya oblast on typical medium loamy sierozem irrigated

long before against the background of P60 in the form of

Card 1/2 Tsentral maya stanting redobreniy i agropochwovedeniza) Usesoyumnoo maurhou inslad, in ta behlopkovorlatva

USSR/Cultivated Plants - Commercial. Oil-Bearing. Sugar-Bearing. M Abs Jour : Ref Zhur Biol., No 18, 1958, 82429

P_C. After application in the form of top dressing, ammonium bicarbonate was equal in effectiveness to N_{aa}. Under the conditions of the hot climate of Central Asia, this fertilizer decomposes rapidly in storage (even under the conditions of good storage in fitted out warehouses the losses for 4 months reach 32%). — A.M. Smirnov

Card 2/2

- 84 ..

YAROVENKO, G. I., Cand Agri Sci — (diss) "The effect of the form of nitrogen fertilizers on the yield of cotton," Moscow, 1958, 19 pp, 150 cop. (Sci Res Institute for Fertilizers and Insectofungicides im Prof. Ya. V. Samoylov) (KL, 45-60, 127)

J . USSR COUNTRY : Soil Science. Mineral Fertilizers. CATEGORY : 5ZhBiol., No. 2 3 1958, No. 104469 ASS. JOUR. : Protasov, P. V.; Yarovenko, G. I. AUTHOR : The Role of Antiseptics in Increasing the Effectiveness of INST. Nitrogen Fertilizers on Irrigated Cotton Fields TITLE : Udobreniye i urozhay, 1958, No. 2, 31-34 ORIG. PUL. : Field experiments carried out on cotton-growing collective farms of Uzbekistan (the soil is typical sierozem with ARCTRACT long-standing cotton culture) showed that the simultaneous introduction under plowing of N_{an} and an antiseptic (N_{ta}) -lindane , paraform, and grandwan -- sharply inhibited the viability of mitrifying and denitrifying bacteria, thus eliminating the possibility of N loss through denitrification and wash-out of N nitrate by autumn-winter pracipitation. Trus, the introduction under plowing of 30 kg/hecters of N + 60 kg/hecters of antiseptic (12% lindane dust, paraform or granoson) increased the growth of cotton, the 1/5 Card:

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001962210016-3"

CATEGORY

ABS. JOUR. : RZhBiol., No. 23 1958, No. 104469

AUTHOR :
INST. :
TLILS :

ORIG. PUB.

ABSTILICT

number of bolls on the plants, and the yield of cotton wool by 3-4 centners/hectars in comparison with Nna alone, introduced in the same dose under plowing or as a top dressing. The application of lindane dust produced the greatest effect. Nts, applied under plowing at the rate of 30 kg/hectare instead of Nna, showed analogous action. Similar results were obtained in another field emeriment. Laboratory investigations (experiment in Petri dishes) that the addition to Nna or Nna of small doses of lindane and Nts dust was accompanied by a considerable accumulation of ammonium Nand by its slower oxidation to nitrates.—0. P. Medvedeva

Card:

2/2

CAUSIPY : Sality:s: . Lants. Councerald. Oldforms. CATTOORT Super-Berring. : RZhBiol., No. 4, 1959, Mn. 15732 ABS. JOUR. : Bragasov, f.; Tanavata, G. Cotton Growing Research Inst., Uzbek SSR AUTHOR INSP. Effectiveness of Presowing Placement of Ammonia TITE Sulfate under Cotton. burg, gus. : Halapkovodetvo, 1958, No.2, 31-36 . Findings of experiments of the central station AFSTRACT of fertilizers and agricultural soil science of cotton growing research institute the of Uzbek SSR, and also an experiment of the agricultural chemistry laboratory of the Chinazskaya MTS on the advantage of Na as compared with Nan in case of presowing placement of Na under cotton (in sierozems) at a rate of 25 to 30 % of the annual quota. Organizational economic advantages of this method are also indicated. -- B. L. Hlycchks-furvion CARD:

115

E ARPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001962210016-3'

DATEGORY

Cultivated Plants. Cornercial. Oleiferous.

ES. JOUR.

Curar-Bearing, 1959, No. 15731

JUTHOR

NST. ITLE

Protesov P.V. Veroveni G.I. Presowing Placement of Liquid Nitrogen Ferti-

lizers under Cotton.

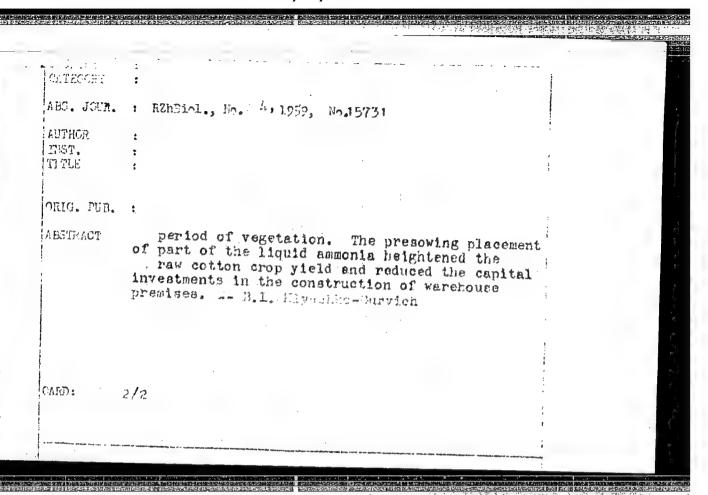
MIG. PUB. : Udobreniye i urozhay, 1958, No.3, 35-38

BSTRACT

: The economic estimates and findings are cited of experiments of the central station of fertilizers and agrosoil science of the All-Union Cotton Scientific Research Institute for 1957, conducted in sierozems in a number of kolkhozy of the Uzbek SSR, on the advantages of placing part of the liquid fertilizers amounting to 30 % of the annual ploughland as compared to using the entire annual quota of liquid ammonia only as supplementary fertilizer in the

ARD:

1/2



YAROVENKO, G.I.

Effect of watering on the efficiency of nitrogen fertilizers. Dokl. AN Uzb. SSR no.3:55-57 158. (MIRA 11:6)

1. TSentral'naya stantsiya udobreniya i agropochevovedeniya Mauchno-issledovatel'skogo khlopkovogo instituta Uzbekskoy akademii sel'sko-khozyaystvennykh nauk. Predstavleno chlenom -korrespondentom AN UzSSR A.M. Mal'tsevym.

(Nitrates) (Fertilizers and manures)

YAROYENKO, G.I.

Wiffect of benzene hexachloride on the effectiveness of ammonium nitrate plowed under before cotton seeding. Dokl. AN UR. SSR (NIRA 11:8), no.5:147-149 158.

YAROTENKO, G.I.

Effect of form of nitrogen fertilizers on the nitrate and ammonium nitrogen in Sierozem soils. Dok. AN UzSSR no.10:41-44 '58. (MIRA 11:12)

1. TSentral nava stantsiya udobreniya i agropochvovedeniya Vsesoyuznogo nauchno-issledovatel skogo instituta khlopkovodstva Akademii sel skokhozyaystvennykh nauk UzSSR. Predstavleno akademikom Akademii sel skokhozyaystvennykh nauk UzSSR S.N.Ryzhovym. (Sierozem soils) (Fertilizers and manures)

YAROVENKO, G.I.

Testing the effect of ammonia water on cotton plantings. Dokl. (MIRA 12:1) AN Uz.SSR no.12:55-57 158.

l. TSentral naya stantsiya udobreniya i agropochvovedeniya Vsesoyuznogo nauchno-issledovatel skogo instituta khlopkovodstva i Akademiya sel'skokhozyaystvennykh nauk UzSSR. Predstavleno deystvitel nym chlenom Akademii sel skokhozyaystvennykh nauk UzSSR N.M. Mannanovym. (Ammonium hydroxide)

(Cotton-Fertilizers and manures)

CIA-RDP86-00513R001962210016-3" **APPROVED FOR RELEASE: 09/01/2001**

YAROVENKO, G. I.

tellist accounting telliness are not required to

Time for applying liquid nitrogen fertilizers to cotton. Dokl.AN Uz.SSR no.3:47-49 59. (MIRA 12:7)

1. TSentral naya stantsiya udobreniya i agropochvovedeniya Vsesoyuznogo nauchno-issledovatel skogo instituta khlopkovodstva. Predstavleno deystvitel nym chlenom Akademii sel skokhozyaystvennykh
nauk UzSSR N.M.Mannanovym.

(Cotton-Fertilizers and manures)

UMAROV, A.A.; YAROVENKO, G.I.

Reflect of armonia and nitrate mutrition of varying waters y. Uzb. yield of cotton under the conditions of varying waters y. Uzb. biol. zhur. 7 no.1:17-19 '63 (MIRA 17:7)

1. Vsesoyuznyy ordena Lenina nauchmo-issledovatel'skiy institut khlopkovodstva.

YAROVENKO, G.I.; UMAROV, A.A.

对行过于22年的双联的证据是是是要实行的对话,可以行为证据的证据是可以是是不多的对象。

Effect of the size of fractions of urea-formaldehyde fertilizers on the biochemical capacity of soils for nitrate accumulation and the yield of cotton. Uzb. biol. zhur. 7 no.6:62-66 '63. (MTRA 17:6)

1. Vsesoyuznyy ordena Lenina nauchno-issledovatel¹skiy institut khlopkovodstva.

YAROVENKO, G.I., KIR, I.E.

Effect of stimulating and phytocidal doses of insecticides on the biological capacity of soils to the nitrate accumulation, development and yield of the cotton plant. Uzb. biol. zhur. 8 no.2:15-17 164. (MIRA 17:9)

1. Vsesoyuznyy nauchno-issledovatel skiy institut khlopkovodstva, Tashkent.

AROVENKO, F.P.

USSR/Cultivated Plants - Technical Oleacoae, Sugar Plants

M-7

Abs Jour : Ref Zhur - Biol., No 1, 1958, No 1690

: I.P. Yarovenko Author : Not Given

Inst : An Attempt to Grow Sugar Cane in Uzbekistan Title

Orig Pub : Sots. s.kh. Uzbekistana, 1956, No 10, 64-67

Abstract : General description of a 10-year attempt to grow sugar

cane in the Khazarbag sovhoze is given. The methods of agrotechny (periods and methods of planting, system of cultivating the soil, fertilizers, irrigation, means of fighting pests and diseases), which permitted the yield to increase from 100 to 160 (in the initial period of application) to 500-

550 c/h during the years 1952-1953.

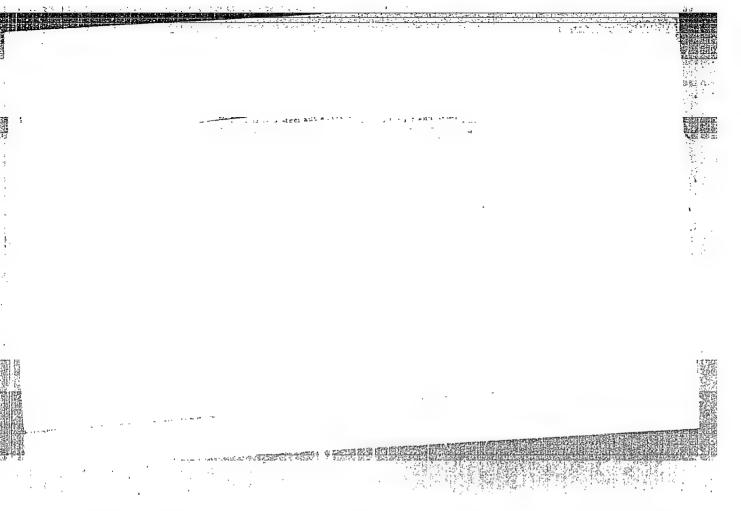
: 1/1 Card

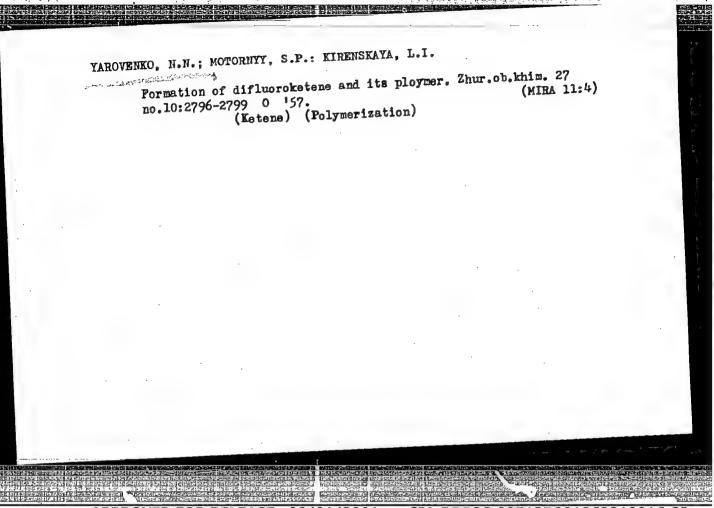
> APPROVED FOR RELEASE: 09/01/2001 redeRIAFRISHED 0513R001962210016-1 redaktor; DUMBRE, I.Ya., tekhnicheskiy redaktor.

[Feldspars; second collection of articles. Translated from the English by A.S. Marfunin Polevye shpaty; 2-i sbornik statei. Perevod s angliiskogo A.S. Marfunina. Pod red. V.P. Petrova. Predisl. V. P. Petrova i A. S. Marfunina. Moskva, Izd-vo inostrannoi (MIRA 9:6) lit-ry. Vol.2. 1956. 366 p. (Feldspar)

YAROVENKO, N.N.; MOTORNYY, S.P.; KIRKNSKAYA, L.I.; VASIL'YEVA, A.S.

Reaction of halide anhydrides of fluorinated carboxylic and
thiocarboxylic acids with sedium azides. Zhur. ob. khim. 27
thiocarboxylic acids with sedium azides. Zhur. ob. khim. 27
(MIRA 1019)
no.8:2243-2246 Ag 157.
(Sodium azide) (Acids, Fatty)





VAROVENKO, N.W.; VASIL'IEVA. A.S.

New means of introducing trihalogen methyl group into organic compounds. Zimr.ob.khim. 28 no.9:2502-2504 S '58. (MIRA 11:11) (Methyl group)

TAROVENKO, N.N.; MOTOHNYY, S.P.

Preparation of M-trifluomethylthiocarbamic acid esters. Zhmr.ob., (MIRA 11:11) (Carbamic acid)

(Carbamic acid)

APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86-00513R001962210016-3"

Properties of organic compounds in the light of the Mendeleev

Periodic Law. Pert 1: Boiling temperature of fluorine compounds

Periodic Law. Pert 1: Boiling temperature of fluorine compounds

and other halides. Zhur.ob.khim. 28 no.9:2506-2509 S '56.

(MIRA 11:11)

(Halogen compounds) (Boiling points)

Yarovenko, N. N., Gaziyeva, G. B., Shemanina, V. N., Fedorova, N. A.

sov/79-29-3-38/61

TITLE:

Syntheses of Organoselenium Compounds Using Carbon Selenide as the

Initial Product (Sintezy selenoorganicaeskika

soyedineniy, iskhodya iz selenougleroda)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 3,

pp 940-942 (USSR)

机舒取利用企业组织的结构 建模型的经验设计 经国际的 网络阿拉特里亚国际的经济管辖的经济区域的 医足迹经的 有过的地名 经金额分配

ABSTRACT:

The aim of the investigations reported in the present paper was the synthesis of new selenium compounds, using carbon selenide as initial product. Carbon selenide is known to be one of the simplest and best accessible selenium carbon compounds. It is formed in the reaction of carbon tetrachloride with phosphorus pentaselenide (Refs 1,2), cadmium selenide (Ref 3) or with hydrogen selenide, as well as in the heating of elementary selenium with methylene chloride in the nitrogen current (Ref 5); the last method is considered the best. Carbon selenide

readily reacts with chlorine under formation of

Card 1/3

trichloromethyl selenium chloride (Ref 5)

CIA-RDP86-00513R001962210016-3" **APPROVED FOR RELEASE: 09/01/2001**

Syntheses of Organoselenium Compounds Using Carbon Seleniac as the Initial Product

507/79-29-3-38/61

Cl₂ CCl₃SeCl. At low temperatures it is possible to obtain higher yields (up to 73%) of trichloromethyl selenium chloride. The authors found that the latter readily reacts with potassium cyanide under formation of trichloromethyl selenium cyanate: CCl₃SeCl CCl₃SeCN. In the reaction of trichloromethyl selenium chloride with ethylene trichloromethyl-β-chloroethyl selenide is formed:

 $\begin{array}{c} \text{CH}_2\text{=}\text{CH}_2\text{=}\text{CH}_2\text{CCl}_3\text{SeCH}_2\text{CH}_2\text{Cl. In the reduction of}}\\ \text{CCl}_3\text{SeCl} &\longrightarrow^2\text{CCl}_3\text{SeCH}_2\text{CH}_2\text{Cl. In the reduction of}}\\ \text{trichloromethyl selenium chloride with metallic tin in the}\\ \text{trichloromethyl selenium the dimer of the selenium carbonyl}\\ \text{hydrochloric acid medium the dimer of the selenium carbonyl}\\ \text{chloride is obtained: CCl}_3\text{SeCl} &\xrightarrow{\text{Sn}}_{\text{CCl}_2\text{Se}})_2\text{. In the} \end{array}$

Card 2/3

reaction of carbon selenide with selenium dioxide the

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001962210016-3

Syntheses of Organoselenium Compounds Using Carbon

507/79-29-3-38/61

Selectide as the Initial Product

carbon selenium oxide is formed: $CSe_2 \xrightarrow{SeO_2 + oleum} CSeO$.

There are 5 references.

SUBMITTED:

February 7, 1958

Card 3/3

CIA-RDP86-00513R001962210016-3" APPROVED FOR RELEASE: 09/01/2001

507/79-29-7-12/83

5(3) AUTHORS: Motornyy, S. P., Kirenskaya, L. I., Yarovenko, K. N.

TITLE:

New N-Trifluoromethyl Carbaminates

(Novyye efiry N-triftormetilkarbaminovoy kisloty)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 7, pp 2157-2159 (USSR)

ABSTRACT:

According to data from publications fluorinated alkyl isocyanates show a high reactivity (Ref 1). They enter especially easily reaction with alcohols and phenols to form esters of M-perfluoro alkyl carbamic acid, c.C.

 $_{R_{F}NCO} \xrightarrow{CH_{3}O\Pi} _{R_{F}NHCO_{2}CH_{3}}$

In papers published earlier by the authors (Ref 2) the reactions of trifluoromethyl isocyanate with halogen hydracid and mercaptans were described. Since the investigation of the chemical properties of alkyl isocyanates and their fluorinated derivatives is of certain interest, the present paper deals with the synthesis of some new N-trifluoromethyl carbaminates. The constants and analytical data of the new compounds are tabulated.

Card 1/2

New N-Trifluoromethyl Carbaminates

807/79-29-7-12/83

Trifluoromethyl isocyanate reacts with normal alcohols under strong heating. For this reason the reaction of the equimolecular amounts of trifluoromethyl isocyanate and alcohol was carried out in a closed glass ampoulo with intense cooling. Yields were 55 to (in individual cases) 70-85 %. More details are given in the experimental part. There are 1 table and 2 Soviet references.

SUBMITTED:

June 6, 1958

Card 2/2

5(3) AUTHORS:

Yarovenko, H. N., Raksha, M. A.

507/79-29-7-13/83

TITLE:

Fluorination by Means of &-Fluorinated Amines

(Ftorirovaniye's pomoshch'yu oftorirovannykh aminov)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 7, pp 2159-2163 (USSR)

ABSTRACT:

In the investigation of the chemical properties of ∞-fluorinated amines the authors succeeded in synthesizing new organofluorine compounds. They found that the amines RCF_NR' react with alcohols to form alkyl fluorides. This reaction takes place easily with a simple mixing (yield up to

66 %)(Scheme 1). The reactions of the fluorinated amines RCF2NR2 were carried out in a similar way with carboxylic acids,

with their salts or with thiocarboxylic acids under the formation of acid fluorides of carboxylic acids (Scheme 2). According to references 1 and 2, also the reactions of ∞-fluorinated amines take place with H₂S and H₂Se under the formation of the hitherto unknown dialkyl amides of fluorinated thiocarboxylic and

selenium carboxylic acids:

Card 1/2

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CIA-RDP86-00513R001962210016-3"

Fluorination by Means of &-Fluorinated Amines

507/79-29-7-13/83

 $chrclcf_2 N(c_2 H_5)_2 \xrightarrow{H_2 S} chrclcsN(c_2 H_5)_2$

 $\begin{array}{ccc} \text{CHF}_2\text{CF}_2\text{N}(\text{C}_2\text{H}_5)_2 & \longrightarrow & \text{CHF}_2\text{CSN}(\text{C}_2\text{H}_5)_2 \\ \text{CHF}_2\text{CF}_2\text{N}(\text{C}_2\text{H}_5)_2 & \longrightarrow & \text{CHF}_2\text{CSeN}(\text{C}_2\text{H}_5)_2 \end{array}$

The high mobility of fluorine atoms in &-position to nitrogen and the ease with which they may be replaced by elements of group 6 may be explained by po-conjunction (Ref 4). The fluorinated amines necessary for these reactions are obtained by the reaction of the secondary amines with fluorinated olefines (Refs 2, 3) which takes place especially easily in the case of addition of secondary amines to trifluoro chloroethylene (Scheme 4). There are 5 references, 2 of which are Soviet.

SUBMITTED:

June 6, 1958

Card 2/2

CIA-RDP86-00513R001962210016-3" APPROVED FOR RELEASE: 09/01/2001

SOV/79-29-7-14/83

5(3)
AUTHORS:

Yarovenko, N. N., Motornyy, S. P., Vasil'yeva, A. S.,

Gershzon, T. P.

TITLE:

Difluoro Chloromethyl Sulphene Chloride

(Diftorkhlormetilsul'fenkhlorid)

PERIODICAL:

Zhurnal obshchey khimii, 1959; Vol 29, Nr 7, pp 2163-2165 (USSR)

ABSTRACT:

The purpose of the present paper was the synthesis of the above compound. In contrast to trichloro methyl sulphene chloride, the product of its reaction with diethyl amine, trichloro methyl—(N-diethyl)-sulphene amide, reacts with antimony trifluoride in the presence of small amounts of SbCl₅, without separation

of the C-S bond, to form fluorodichloro-, difluorochloro-, and probably trifluoromethyl-(N-diethyl)-sulphene amides. In this connection heating and its duration play an important part. Below 65° practically only fluoro dichloromethyl-(N-diethyl)-sulphene amide is formed. At 67° and after heating during 1 1/2 hours difluoro chloromethyl-(N-diethyl)-sulphene amide (25 %) is formed in the mixture with fluoro dichloro- and trichloro methyl-(N-diethyl)-sulphene amide. Since difluoro chloromethyl-(N-diethyl)-sulphene amide is very unstable, it is not necessary

Card 1/2

Difluoro Chloromethyl Sulphene Chloride

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to separate it from the reaction mass. The liquid must only be separated from the solid, resinous reaction products and then saturated with dry HCl (Ref 4)(Scheme 3). The thus obtained mixture of trichloro-, difluoro chloro-, and fluorodichloro methyl sulphene chloride may easily be separated by distillation in a column. The effect of temperature and the duration of heating on the yield of difluorochloro- and fluorodichloromethyl sulphene chlorides may be seen from a table. There are 1 table and 4 references, 1 of which is Soviet.

SUBMITTED:

June 6, 1958

Card 2/2

YAROVENKO, N.N.; RAKSHA, M.A.; SHEMANINA, V.N.

SERVICE THE SERVICE OF THE SERVICE O

Synthesis of halogenated bis (alkyl) diselenides and symmetrical bis (chlorodifluoromethyl) disulfide. Zhur. ob. khim. 30 no.12:4069-(MIRA 13:12) 4071 D 160.

(Diselenide)

(Disulfide)

YAROVENKO, N.N.; MOTORNYY, S.P.

Methods of synthesizing fluorine analogues of dichloroformoxime.

Zhur. ob khim. 30 no.12:4066-4069 D '60. (MIRA 13:12)

(Formaldehyde)

87535 \$\079\60\030\012\020\027 8001\B064

5.3600

Yarovenko, N. N. and Raksha, M. A.

TITLE:

AUTHORS:

Synthesis of Tetrafluoro Dimethyl-diselenide and Some of Its

Properties

PERIODICAL: Zhurnal obshchey khimii, 1960, Vol. 30, No. 12, pp. 4064-4066

TEXT: The authors found already in a previous paper (Ref.1) that - analogously to hexafluoro dimethyl-diselenide - the tetrafluoro dimethyl-diselenide

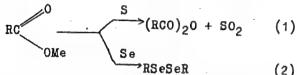
 $(CF_3CO_2)_2Hg \xrightarrow{S} CF_3SeSeCF_3$ $(CHF_2CO_2)_2Hg \xrightarrow{Se} CHF_2SeSeCHF_2$

is formed in the decarboxylation of the mercury salt of difluoro acetic acid in the presence of selenium. A comparison of these reactions with the well-known reaction of trifluoro acetic acid salts with sulfur (Ref.2) well-known that the decarboxylation of fluoro carboxylic acid salts in the shows that the decarboxylation of fluoro carboxylic acid salts in the presence of the elements of group VI may proceed in two directions. The reaction depends on the position of the salt former in the periodic table;

Card 1/2

X

Synthesis of Tetrafluoro Dimethyl-diselenide 5/079/60/030/012/020/027 and Some of Its Properties B001/B064



Apparently, also at a further rise of the atomic number of the element, a decomposition of the salts in direction (2) occurs. It may be assumed that in the reaction of tellurium with salts of fluorinated carboxylic acids, fluorinated dialkyl ditellurides are formed. Difluoro methyl selenium bromide CHF₂SeBr resulted from the reaction of tetrafluoro

dimethyl-diselenide with bromine. It reacts readily with calcium cyanide under the formation of difluoro methyl selenocyanide and adds to ethylene under the formation of difluoro methyl- β -bromoethyl selenide:

There are 3 references: 2 Soviet and 1 British. SUBMITTED: January 11, 1960 Card 2/2

87536

S/079/60/030/012/021/027 B001/B064

5.3600

AUTHORS: Yarovenko, N. N., Raksha, M. A., and Shemanina, V. N.

TITLE: Synthesis of Halogenated Dialkyl Diselenide and the Symmetrical Tetrafluoro Dichloro Dimethyl Disulfide

PERIODICAL: Zhurnal obshchey khimii, 1960, Vol. 30, No. 12, pp. 4069 - 4071

TEXT: Considering the papers of Refs.1-5 on the synthesis of the halogenated dialkyl selenides, the authors found that the fluorinated dialkyl diselenides are also obtained when monoselenium bromide is reacted with tetrafluoro ethylene:

to temperature and longer heating. The best diselenide yield is obtained by gradually heating the initial products to 160°C in an inert solvent. When the reaction mixture is rapidly heated to a high temperature, the monoselenium bromide brominates the diselenide under the separation of

Card 1/2

1

Synthesis of Halogenated Dialkyl Diselenide and the Symmetrical Tetrafluoro Dichloro Dimethyl Disulfide

87536 \$/079/60/030/012/021/027 B001/B064

considerable amounts of elementary selenium. The structure of the diselenide obtained was confirmed by a chlorination to 2-bromo-1,1-2,2-tetrafluoro ethyl selenium chloride (BrCF₂CF₂SeCl). Some halogenated alkyl selenium halides may be reduced to halogenated dialkyl diselenides (CF₃SeSeCF₃). 2,2'-dichloro diethyl diselenide may be obtained by reacting 2,2'-dihydroxy diethyl diselenide with concentrated hydrochloric acid. The initial product for this reaction was obtained by reacting ethylene oxide with H₂Se under pressure. The dialkyl diselenides obtained are colored, bad smelling liquids which are insoluble in water. There are 8 references: 4 Soviet, 3 US, and 1 British.

SUBMITTED: January 28, 1960

Card 2/2

27504 s/079/61/031/009/005/012 D215/D306

5.3600

Yarovenko, N.N., and Vasil'eva, A.S. AUTHORS:

Dichloroperfluorodivinylsulphide and sulphides

with monofluorochloroethyl group TITLE:

PERIODICAL: Zhurnal obshchey khimii, v. 31, no. 9, 1961, 3021 - 3023

TEXT: The work was conducted to establish the order of addition of sulphur monochloride and hydrogen sulphide to fluorinated olefines under pressure and the action of light. It has been established that when a mixture of hydrogen sulphide and trifluorochloroethylene is irradiated in a sealed ampoule, in the presence of benzoyl peroxide, dichloroperfluorodivinylsulphide and its poly-

 $\xrightarrow{\text{CFCl}} [\text{S(CF}_2 - \text{CHFCl})_2] \xrightarrow{\text{HF}} \text{S(CF} = \text{CFCl})_2 + [\text{S(CF} = \text{CFCl})_2 + \text{CFCl}]_2$ mers are obtained, $= cFcl)_2$ _n.

Card 1/4

CIA-RDP86-00513R001962210016-3 **APPROVED FOR RELEASE: 09/01/2001**

s/079/61/031/009/005/012 D215/D306

Dichloroperfluorodivinylsulphide ...

When sulphur monochloride and vinyl fluoride are reacted under similar conditions, 2,2'-difluoro-2,2'-dichlorodiethylsulphide is formed

 $s_2cl_2 \xrightarrow{CH_2 = CHF} s(cH_2cHFcl)_2 + S.$

The structure of this compound is confirmed by the inertness of all C-Cl and C-F bonds. Prolonged stirring of the compound in water at room temperature fails to produce ionic fluorine or chlorine. In compounds with one 2-chloroethyl group and one 2'-fluoro-2'-chloro- or 2,2'-difluoroethyl group only one chlorine atom of 2-chloroethyl group is easily hydrolized. These compounds were prepared by reacting 1-fluoro-1-chloro-2-bromoethane, 1-fluoro-1,2-dichloroethane and 1,1-difluoro-2-bromoethane with sodium 2hydroxyethylmercaptide followed by substitution of the hydroxyl group with chlorine CII2CIIFCI

CH2CHFCI SOCI. SC HOCH2CH2SNa CH3BrCHFCI SCH2CH3OH

Card 2/4

27504 S/079/61/031/009/005/012 D215/D306

Dichloroperfluorodivinylsulphide ...

The order of addition of sulphur monochloride to vinyl fluoride is confirmed indirectly by the fact that when sulphur monochloride is reacted with vinyl chloride 2,2,2',2'-tetrachlorodiethylsulphide is produced and the latter hydrolyzes in water to form dialdehyde

$$S_2Cl_2 \xrightarrow{CH_3=CHCl} S(CH_2CHCl_2)_2 \xrightarrow{H_1O} S\left(CH_2C \swarrow_{II}^O\right)_2$$

Preparation of 2,2'-difluoro-2,2'-dichlorodiethylsulphide involved sealing 20.3 g of $S_2\text{Cl}_2$, 18.5 g of vinyl fluoride and 0.2 g of benzoyl peroxide in an ampoule and irradiating the mixture with a 500 W lamp for 200 hrs. Vacuum distillation yieled 9 g of fraction b.pt. $78-79^{\circ}\text{C/9}$ mm, n_D^{17} - 1.4813, d_4^{17} - 1.4550, corresponding to the formula $C^4\text{H}_6\text{SF}_2\text{Cl}_2$. 2,2'-difluoro-2,2'-dichlorodiethylsulphine-p-toluenesulphonylimine m.pt. 139°C corresponding to the formula Card 3/4

Dichloroperfluorodivinylsulphide ...

27504 S/079/61/031/009/005/012 D215/D306

CH₃C₆H₄SO₂NS(CH₂CHFCl)₂ was prepared by shaking 0.02 q.mol. 2,2'-difluoro-2,2'-dichlorodiethylsulphide with CH₃C₆H₄SO₂NNaCl. 3H₂O solution for 1 hr. and recrystallization from alcohol. 2,2,2'2'-tetrachlorodiethylsulphide was prepared by irradiation of a mixture of 0.2 q. mol. S_2 Cl₂. 0.2 g benzoyl peroxide and 0.2 g mol. vinylchloride for 15 days. Vacuum distillation yielded 36 % C₂H₆SCl₄ b.pt. 106° C/8mm, 106° C/8mm, 106° C/8mm, 106° C/16 mm, 106° C/16 mm,

YAROVENKO, N.N.; RAKSHA, M.A.; GAZIYEVA, G.B.

New methods for the preparation of esters and selenious acid ester halides. Zhur.ob.khim. 31 no.12:4006-4010 D '61.

(MIRA 15:2)

(Selenious acid)

TAROVENKO, N.N.; RAKSHA, M.A.

Reaction of diffuoromethyldifluoroacetate with potassium fluoride.

Zhur.ob.khim. 31 no.12:4011-4012 D '61. (HIRA 15:2)

(Acetic acid)

(Potassium fluoride)

RAKSHA, M.A.; YAROVENKO, N.N.

Reaction of difluoroacetates with arsenic, arsenic trichloride, and nitrosyl chloride. Zhur. ob. khim. 32 no.1:273-274 Ja '62.

(MIRA 15:2)

(Acetic acid) (Arsenic chloride)
(Nitrosyl chloride)

YAROVENKO, N.N., doktor khimicheskikh nauk

Psychotomimetic agents. Zhur. VKHO 9 no.4:448-455 '64.

(MIRA 17:19)

YAROVENKO, O.; MIROSHNIK, A.

Use of diffusion screens in rotary apparatus. Sakh. prom.
(MIRA 16:8) 37 no.8:71 Ag 163.

1. Glavnyy inzh. Krasnyanskogo sakharnogo zavoda (for Yarovenko).
2. Glavnyy tekhnolog Krasnyanskogo sakharnogo zavoda (for Miroshnik).

(Diffusers)

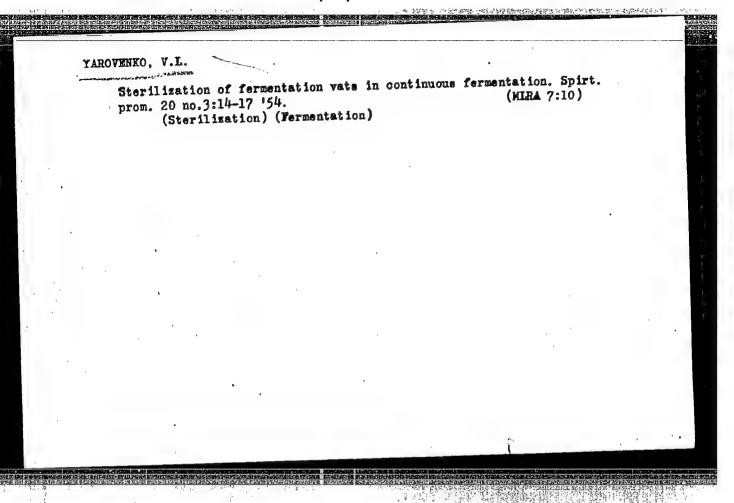
SILIN, P.M.; LITVAK, I.M.; BARABANOV, M.I.; LIKHITSKIY, M.Kh.;
BODNAR, S.G.; ROSTRIPENKO, I.A.; SOFRONYUK, L.P.;
YAROVENKO, O.A.; MIROSHNIK, A.P.; IVASENKO, G.

Accelerating the sedimentation in settlers. Sakh. prom. 36 no.7:9-17 J1 '62. (MIRA 17:1)

1. Moskovskiy tekhnologicheskiy institut pishchevoy promyshlennosti (for Silin). 2. Kiyevskiy tekhnologicheskiy institut pishchevoy promyshlennosti imeni Mikoyana (for Litvak, Barabanov, Likhitskiy). 3. Lannovskiy sakharnyy zavod (for Bondar¹, Ivasenko). 4. 2-y im. Petrovskogo sakharnyy zavod (for Rostripenko). 5. Gindeshtskiy sakharnyy zavod (for Sofronyuk). 5. Krasnyanskiy sakharnyy zavod (for Yarovenko, Miroshnik).

- 1. YAROVENKO, V., BAYKO, V. P.
- 2. USSR (600)
- 4. Soils-Analysis
- 7. Problem of early spring tillage of the soil. Pochwovedenie. No. 10, 1952.

9. Monthly List of Russian Accessions, Library of Congress, January 1953, Unclassified.



YAROVENKO, V.L.

Investigation of the movement processes of liquid in a fermentation (MLRA 8:10)

vessel and battery. Spirt.prom.21 no.2:6-10 155.

1. Vsesoyuznyy nauchno-issledovatel'skiy institut sakharnoy promyshlennosti.

(Fermentation)

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001962210016-3

YNROVENKO, V.L.

USSR/Chemical Technology - Chemical Products and Their Application. Fermentation Industry, I-27

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 63542

Author: Yarovenko, V. L.

Institution: None A-U Sciles Inst alcohol Industry

Title: Effect of Conditions on Displacement of Liquid in the Vessels of a Fermentation Battery

Original
Periodical: Spirt. prom-st', 1955, No 3, 15-19

Abstract: Effectuation of continuous fermentation of starchy raw materials necessitated the carrying out of laboratory experiments to determine the dynamics of movement of a sugar solution in vessels connected in series. Described are the experimental techniques and the results obtained.

Card 1/1

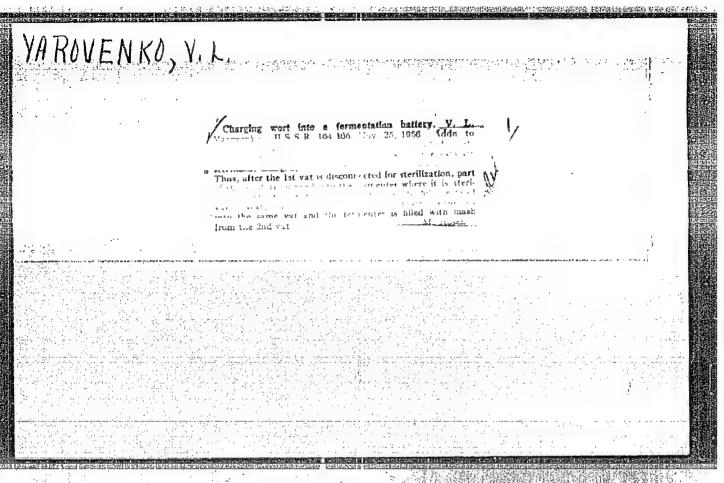
"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001962210016-3

YAROVENKO, V. L.

YAROVENKO, V. L. "A Continuous Method of Fermentation in the Production of Alcohol from Raw Starch." Min Higher Education USSR. Leningrad Technological Inst of the Food Industry. Leningrad, 1956. (Dissertation for the Degree of Candidate in Technical Science)

So: Knizhnaya Letopis', No. 19, 1956.

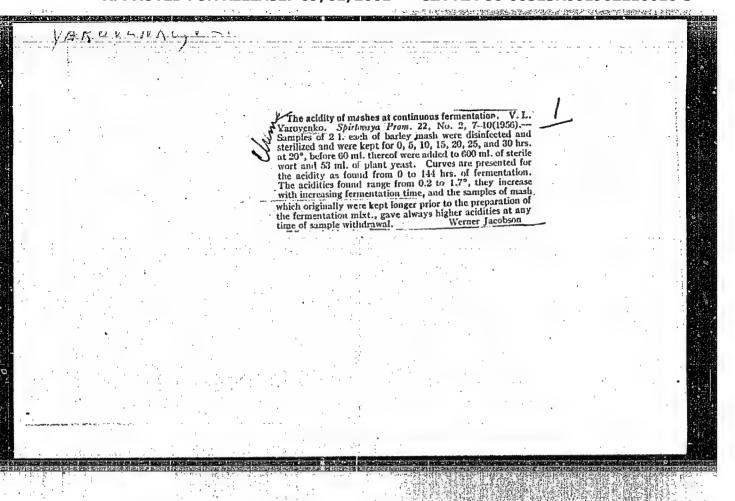


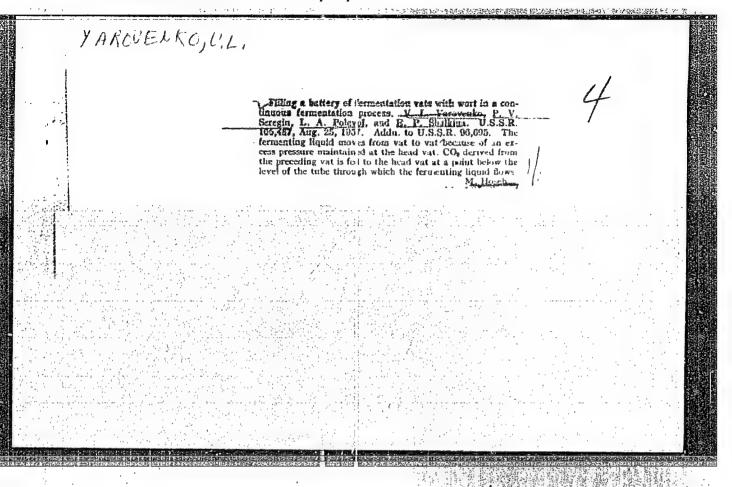
YAROVENKO, V.L.

Plenary session of the scientific council of the All-Union Scientific Research Institute of the Alcohol Industry. Spirt.prom.22 no.1:46 156. (Distilling industries) (MIRA 9:7)

"APPROVED FOR RELEASE: 09/01/2001 CIA-R

CIA-RDP86-00513R001962210016-3





YAROUNTO he V.L.

USSR/General Problems. Methodology. History. Scientific

Institutions and Conferences. Instruction. Questions Concerning Eibliography and Scien-

tific Documentation

Abs Jour : Ref Zhur-Khimiya, No 3, 1958, 6830

Author : V. B. Fremel' and V. L. Yarovenko

Inst : All-Union Scientific Research Institute of

Alcohol and Liqueur-Vodka Industry

Title : Work of All-Union Scientific Research Insti-

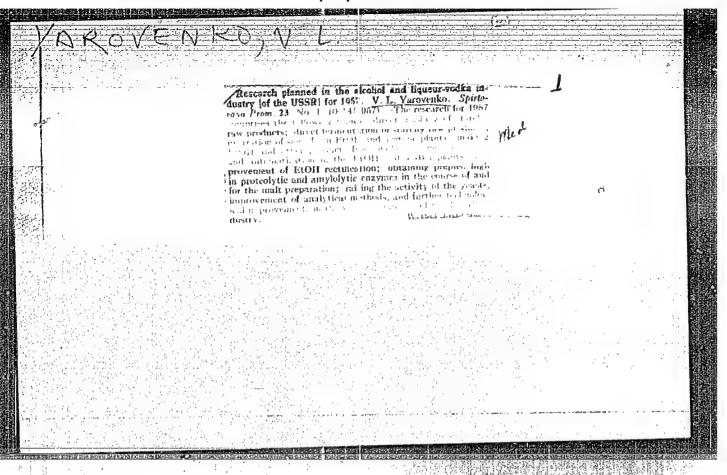
tute of Alcohol and Liqueur-Vodka Industry

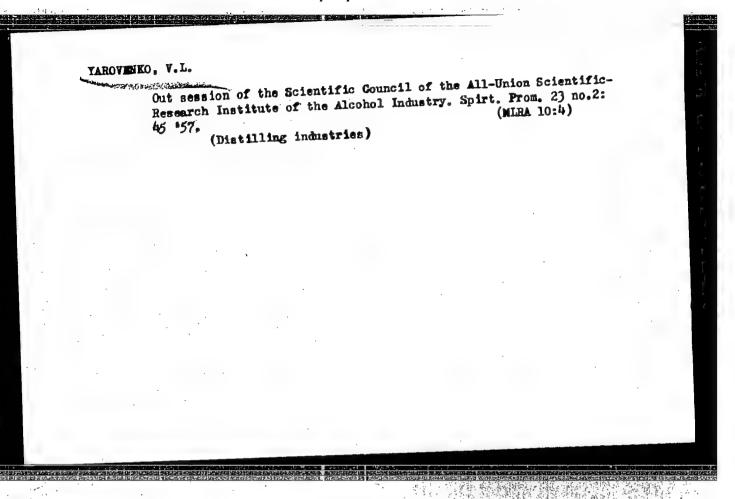
Orig Pub : Spirt. prom-st', 1957, No 7, 18-24

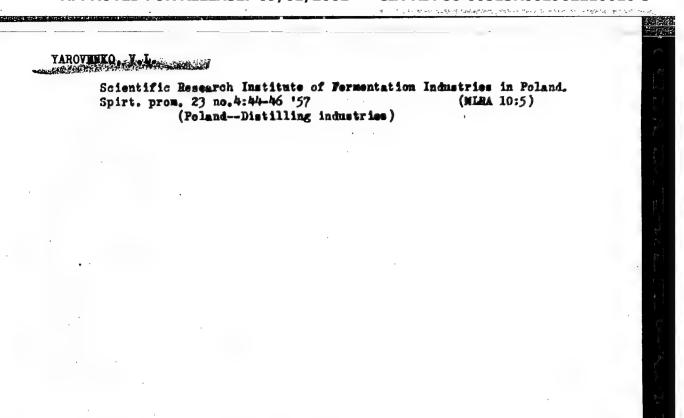
Abstract : To the 40th anniversary of the Great October

Socialist Revolution.

Card 1/1







YAROVENKO, V.I.

Acetone-butyl industry of the Polish People's Republic. Spirt. prom.
(MIRA 10:8)

23 nc.5:14-19 '57.

1. Vsesoyusnyy nauchno-issledcvatel'skiy institut spirtovoy promyshlornosti.
(Poland--Acetone) (Poland--Butyl alcohol)

"APPROVED FOR RELEASE: 09/01/2001 CIA

CIA-RDP86-00513R001962210016-3

Work of the All-Union Scientific Research Institut of the Alcohol, Liqueur and Vodka Industry. Spirt.prom. 23 no.?:
18-24 157. (Distilling industries)

"APPROVED FOR RELEASE: 09/01/2001 CIA-

CIA-RDP86-00513R001962210016-3

YAROVENKO, V.L.; KOMAROV, A.F.

Processing best molasses at alochol plants in Czechoslovakia.

Processing best molasses at alochol plants in Czechoslovakia.

Spirt.prom. 23 no.8:25-29 157.

(Czechoslovakia--Alcohol)

YAROVENKO, Viktor L'vovich; KUZNETSOV, N.M., retsenzent; MALCHENKO, A.L., spetsred.; KOVALEYSKAYA, A.I., red.; TARASOVA, N.M., tekhn.red.

[Continuous alcohol fermentation] Potochnyi metod spirtovogo brozheniia. Moskva, Pishchepromizdat, 1958. 127 p. (MIRA 12:4) (Fermentation)

YAROVENKO, V.L.; SKALKINA, Ye.P.; PYKHOVA, S.V.; LAZAREVA, A.N.

Continuous fermentation. Trudy TSNIISP no.6:3-2 '52. (MIRA 14:12) (Fermentation)

YAROVENKO, V.L.; SKALKINA, Ye.P.; PYKHOVA, S.V.; LAZAREVA, A.N.

Cyclic semicontinuous fermentation. Trudy TSNIISP no.6:9-14 '38.

(MIRA 14:12)

(Fermentation)

"APPROVED FOR RELEASE: 09/01/2001 CIA-RDP86

CIA-RDP86-00513R001962210016-3

YAROVENKO, V.L.

KOMAROV, A.F.; YAROVENKO, V.L.

Power engineering and mechanization in alcohol plants in

Czechoslovakia. Spirt. prom. 24 no.1:17-23 158. (MIRA 11:3)

(Czechoslovakia.—Distilling industries)

(Czechoslovakia.—Distilling industries)

YAROVENKO, V.L.; NAKHMANOVICH, B.M.; SHCHEBLYKINA, N.A.; SHCHEBLYKIN, N.P.

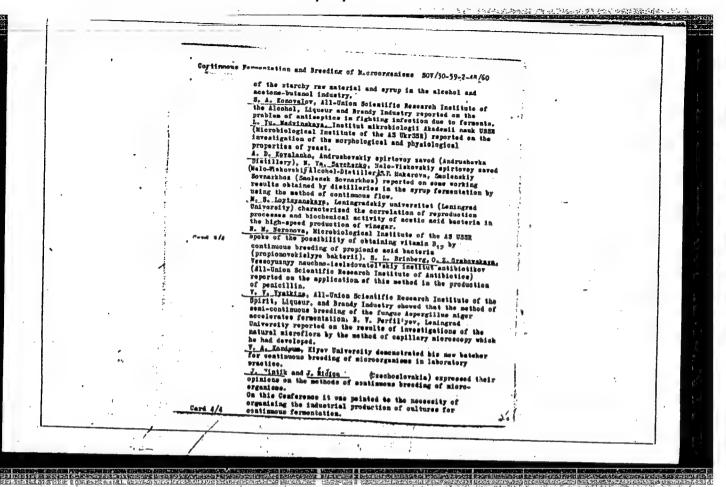
Analysis of the battery method of formenting in the manufacture of butyl alcohol and acetome. Spirt. prom. 24 no.5:5-11 '58.

(Butyl alcohol) (Acetone) (MIRA 11:9)

YAROYENKO, V.L.: MAMUNYA, A.U.

Sterilization of the fermentation battery for the production of alcohol from molasses. Spirt. prom. 24 no.6:10-13 *58., (Fermentation) (MIRA 11:10)

The Institut mikrobiologic Akademic mank SISM (Microbiological Institute of the Academy of Solice SISM) convenued a conference from Guider 13 to 15, 1978 which dealt with the conference from Guider 13 to 15, 1978 which dealt with the investigation of some working rule in this fitted as well as investigation of some working rule in the fitted as well as with the discussion of a further intensification of the with the discussion of a further intensification of the productions healing on the sativity of micropromises. The conference was attunctly more than 200 representatives of academic and microfilic branch crack institutes, enterprises, sowenthouse, universities, as well as fereign scientists. The following lectures were beautiful to the several section of the present of the service of the	
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YAROVENKO, V.L.; SKALKINA, Ye.P.; PYKHOVA, S.V.; LAZAREVA, A.N.

Experience in introducing and developing the continuous method of fermentation in the processing of starchy raw materials.

Trudy TSNIISP no.7:3-16 159. (MIRA 13:9)

(Fermentation) (Alcohol)

SKALKINA, Ye.P.; YAROVENKO, V.L.; PYKHOVA, S.V.; LAZAREVA, A.N.

Multiplication of yeast cells and their distribution in the battery in a continuous fermentation process. Trudy TSNIISP no.7:16-23 '59. (MIRA 13:9)

(Yeast) (Fermentation)

PYKHOVA, S. V.; YAROVENKO, V. L.; SKALKINA, Ye.P.; LAZAREVA, A.N.

Use of the ether - aldehyde fraction as an antiseptic in the manufacture of alcohol. Trudy TSNIISP no.7:25-28 59.

(MIRA 13:9)

(Alcohol) (Antiseptics)

YAROVENKO, V.L.; KOPYLOVA, A.M.

Improved design of a pump for transferring beer. Trudy TSNIISP no. 8:157-164 '59. (MIRA 14:1)

(Distilling industries—Equipment and supplies)

(Pumping machinery)

YAROVENKO, V.L.

All-Union Scientific Research Institute of the Alcohol, Liqueur and Vodka Industries and the 21st Congress of the CPSU. Spirt. prom. 25 no.1:15-16 '59. (MIRA 12:2) (Distilling industries)

YAROVENKO, V.L.

New trends in the scientific research of the All-Union Scientific Research Institute of the Alcohol Industry. Spirt. prom. 25 no.6:11-15 '59. (MIRA 12:12)

(Distilling industries--Equipment and supplies)

"APPROVED FOR RELEASE: 09/01/2001

CIA-RDP86-00513R001962210016-3

ECHOVALOV, S.A.; YAROVENKO, V.L.; BUROVA, M.V.; BOROIKINA, V.V.

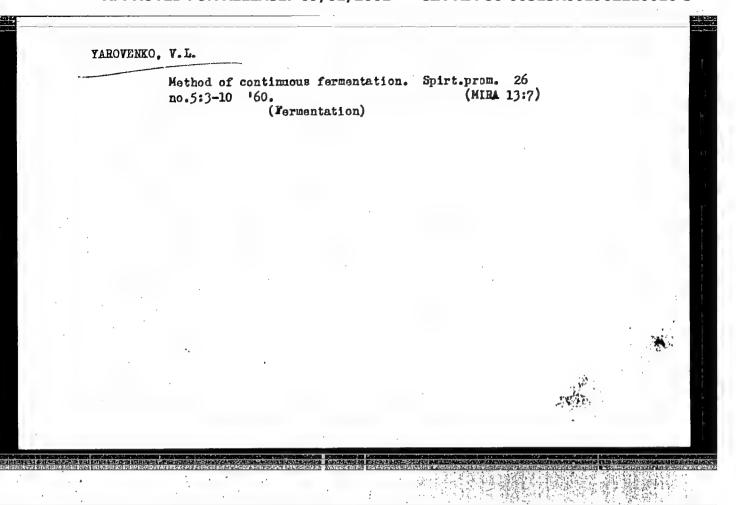
Disinfection of green malt. Spirt.prom. 26 no.1:13-16
(60. (MIRA 13:6)

(Malt--Disinfection)

YAROVENKO, V.L.; SKALKINA, Yo.P.; PYKHOVA, S.V. Combined processing of potatoes into alcohol and starch.

Spirit.prom. 26 no.4:4-7 '60. (MIRA 13:8)

(Potatoes) (Alcohol) (Starch)



YAROVENKO, V.L.; NAKHMANOVICH, B.M.; SENKEVICH, V.V.

Theory of the continuous acetone - butyl alcohol fermentation.

Spirt.prom. 26 no.6:6-9 '60. (MIRA 13:11)

(Fermentation)

YAROVENKO, V.L.; NAKHMANOVICH, B.M.; SHCHEBLYKIN, N.P.; SENKEVICH, V.V.

Study of continuous acetone-butyl fermentation caused by Clostridium acetobutylicum. Mikrobiologiia 29 no. 4:581-586 Jl-Ag '60. (MIRA 13:10)

l. Vsesoyuznyy nauchno-issledovatel'skiy institut spirtovoy promyshlennosti.
(CLOSTRIDIUM ACETOBUTYLICUM)

NAKHMANOVICH, B.M.; SENKEVICH, V.V.; TAROVENKO, V.L.

Use of butyl bacteria for the fermentation of nonedible raw material.

Spirt.prom. 27 no.1:22-25 '61. (MIRA 14:2)

(Fermentation)

YAROVENKO, V. L.

Second International Symposium on Continuous Culture of Micro-Organisms. Spirt. prom. 28 no.8:13-16 '62. (MIRA 16:1)

1. TSentral'nyy nauchno-issladovatel'skiy institut spirtovoy promyshlennosti.

(Fermentation) (Microbiology—Congresses)

YAROVENKO, V.L.; NAKHMANOVICH, B.M.; SENKEVICH, V.V.; SHCHEBLYKIN, N.P.

Continuous acetone-butyl fermentation with an extended battery charging cycle. Izv.vys.ucheb.zav.; pishch.tekh. 2:98-104 '62.

(MIRA 15:5)

1. TSentral'nyy nauchno-issledovatel'skiy institut spirtovoy i likerovodochnoy promyshlennosti.

(Fermentation) (Acetone) (Butyl)

DANILOV, K.G.; YAROVENKO, V.L.

Comparing the simplest modifications of the top part of fermentstion batteries. Spirt.prom. 29 no.4:8-14 '63. (MIRA 16:5)

1. Universitet druzhby narodov (for Danilov). 2. Vsesoyuznyy nauchno-issledovatel'skiy institut fermentnoy i spirtovoy promyshlennosti (for Yarovenko).

(Fermentation—Equipment and supplies)

YAROVENKO, V.L.; USTINNIKOV, B.A.; PYKHOVA, S.V.; LAZAREVA, A.N.

Testing and improvement of the technological flow sheet for the combined processing of potatoes to starch and alcohol in the combined processing of real results of the starch and alcohol in the Michurinsk Distillery. Trudy TSNIISP no.12:46-50 162.

(MIRA 17:3)

YAROVENKO, V.L.; USTINNIKOV, B.A.; PYKHOVA, S.V.; LAZAREVA, A.N.;
KUCHEROVA, E.A.,

Utilization of the cellular juice of potatoes in the combined production of starch and alcohol. Trudy TSNISP no. 13:3-10

162.

(MIRA 17:5)

YAROVENKO, V.L.

Basic tasks of scientific research in the fermentation and distilling industries. Form i spirt. prom. 30 no.3:3-6 '64. (MIRA 18:2)

1. Vsesoyusnyy nauchno-issledovatel'skiy institut fermentnoy i spirtovoy promyshlennosti.

YAROVENKO, V.L.; PYKHOVA, S.V.; USTINNIKOV, B.A.; LAZAREVA, A.N.; MAKEYEV, D.M.

Fermentative hydrolysis of starch in continuous alcohol fermentation. Ferm.i spirt.prom. 31 no.1:5-10 '65. (MIRA 18:5)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut fermentnoy i spirtovoy promyshlennosti.

YAROVENKO, V.L.; USTINNIKOV, B.A.; LEVCHIK, A.P.; NECHIPORENKO, A.A.

Processing of sugar beets in a mixture with grain and potato raw materials and molasses. Ferm. i spirt. prom. 31 no.6:37-40 '165. (MIRA 18:9)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut fermentnoy i spirtovoy promyshlennosti (for Yarovenko, Ustinnikov).

2. Michurinskiy spirtozavod (for Levchik, Nechiporenko).